### Water Environment Research Foundation (WERF) Bioavailable Phosphorus (BAP) Reports

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#### Water Environment Research Foundation (WERF) Bioavailable Phosphorus (BAP) Reports

- WERF Nutrient Challenge
- Overview of Phosphorus Speciation and Bioavailability
- New WERF BAP Reports
  - "The Bioavailable Phosphorus (BAP) Fraction in Effluent from Advanced Secondary and Tertiary Treatment" (NUTR1R06m)
  - "Mineralization Kinetics of Soluble Phosphorus and Soluble Organic Nitrogen in Advanced Nutrient Removal Effluents" (NUTR1R06p)
- Applications

### Water Environment Research Foundation (WERF) Nutrient Challenge

### www.werf.org



Go to **KNOWLEDGE AREAS: Nutrients** >50 completed and ongoing projects

### **Original Objectives**

Provide science-based solutions and recommendations that:

(1) support utility decisions to use sustainable wastewater nutrient removal technologies to meet various receiving water body requirements and other wastewater treatment goals (e.g., climate change, sustainability, cost-effectiveness, reliability), and

(2) inform regulatory decision making that is moving toward increasingly higher levels of nitrogen and phosphorus removal.

#### WERF's Nutrient Challenge includes work completed or done in partnership with many utilities, consultants, universities, organizations, etc., including:

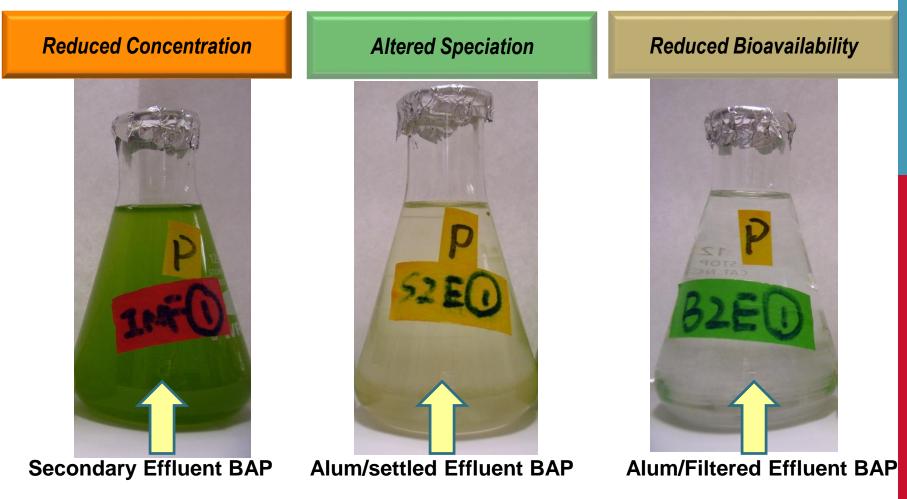
Consulting Firms	Utilities	Other Agencies	Universities
<ul> <li>HDR, Inc. *,+</li> <li>CH2H-Hill +</li> <li>AECOM +</li> <li>Passaro Engineering</li> <li>Black &amp; Veatch</li> <li>Hazen and Sawyer</li> <li>Envirosim, Canada</li> <li>Brown and Caldwell</li> <li>* Lead Contractor + Core team member</li> </ul>	<ul> <li>DC Water, DC</li> <li>Hampton Roads Sanitation District, VA</li> <li>Coeur d'Alene, ID</li> <li>City of Las Vegas, NV</li> <li>Spokane County, WA</li> <li>City of Spokane, WA</li> <li>City of Hayden, ID</li> <li>Inland Empire Paper Co</li> <li>New York City DEP, NY</li> <li>Clean Water Services, OR</li> <li>Spokane County, ID</li> <li>Spokane County, ID</li> <li>Coeur d'Alene, ID</li> <li>City of Las Vegas, NV</li> <li>City of Stamford, CT</li> <li>Alexandria Renew, VA</li> <li>Truckee Meadows, NV</li> <li>Edmonton, Canada</li> <li>Sydney Water Corp.</li> </ul>	<ul> <li>USEPA Region III, Chesapeake Bay Program Scientific Technical Advisory Committee (STAC)</li> <li>Metropolitan Washington Council of Govts (MWCOG)</li> <li>MEGA (Modeling Experts Group of the Americas)</li> <li>Connecticut DEEP / Nitrogen Credit Advisory Board</li> <li>USEPA</li> <li>WEF</li> <li>IWA</li> </ul>	<ul> <li>University of Washington +</li> <li>University of California – Berkeley</li> <li>Columbia University, NY</li> <li>Howard University, DC</li> <li>Gdansk University, Poland</li> <li>Northeastern University</li> <li>Wilfrid Laurier University, Canada</li> <li>University of Waterloo, Canada</li> <li>University of Utah</li> <li>Illinois Institute of Technology</li> </ul>





# **Overview of** Phosphorus Speciation and Bioavailability

### Advanced Treatment and Effluent Nutrient Speciation and Bioavailability



Michael T. Brett & Bo Li Phosphorus Bioavailability Studies, University of Washington

Some Dissolved Organic Nitrogen (DON) and Nonreactive **Dissolved Phosphorus (nRDP) is Readily Available to Algae** and Some is Not

Coconuts and Bananas Postulate -- Bo Li, University of Washington

### **Bioavailable**

- Reactive P (PO<sub>4</sub> <sup>-3</sup>)
- Amino acids

### **Recalcitrants**

#### Inorganic P

- Apatite
- ( Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> )
- FePO₄

### N and P in humic substances Organic P

- Polyphosphate
- AIPO<sub>4</sub> Inositol hexakisphophate
  - L-α-phosphatidyl choline
  - phosphoenol pyruvate
  - Glycerophosphate

WERF Nutrient Webinar, Oct 5, 2011 Dave Stensel, University of Washington

### **Readily Bioavailable**





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Michael T. Brett & Bo Li, University of Washington libo@u.washington.edu

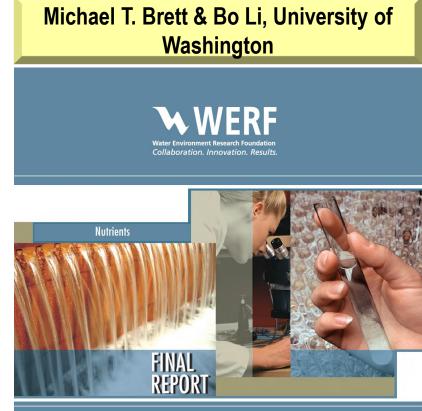




### New WERF Bioavailable Phosphorus (BAP) Reports

#### "The Bioavailable Phosphorus (BAP) Fraction in Effluent from Advanced Secondary and Tertiary Treatment" (NUTR1R06m)

- Algal Bioassays
- Bioavailable P Determined for 17
   Facilities
  - $_{\odot}$  Wide Range of Treatment Processes
    - Chemical and Biological P Removal
    - Single and Two-Stage Tertiary
    - Membranes



The Bioavailable Phosphorus (BAP) Fraction in Effluent from Advanced Secondary and Tertiary Treatment



### **Effluent Sampled**

Bio-P (No Chem)

 Coeur d'Alene MBR
 Snoqualimie
 North Durham

### MBR w/Chem

Broad Run
 Ruidoso

- Single Stage Filter w/Chem
  - $_{\odot}$  Coeur d'Alene TMF
  - Syracuse Actiflo (Onondaga County)
  - DC Water Blue Plains
  - Hayden 1<sup>st</sup> Sage BluePro
  - $_{\odot}$  Las Vegas
  - South Durham
- Dual Stage Tertiary w/Chem
   lowa Hill
  - $_{\circ}$  Coeur d'Alene 2 Stage BluePro
  - Hayden BluePro 2<sup>nd</sup> Stage
  - $_{\circ}$  Farmers Korner
  - o CWS Rock Creek & Durham

### Findings

- Effluent P Speciation and Bioavailability Varies with Treatment Process
  - o >50% Effluent P Recalcitrant in Most Cases
  - Higher Chemical Doses Decreased BAP
- Bioavailability of P Species

Classic SRP Characterization Poor Predictor of BAP

Recalcitrant P Compounds

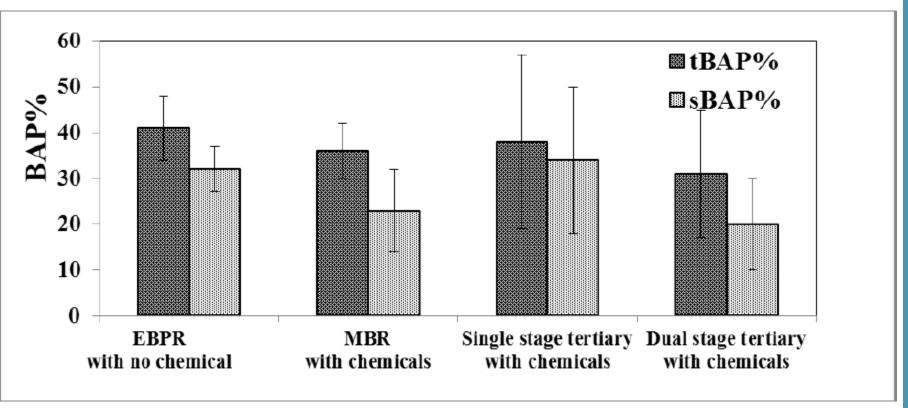
Humic-Metal-P Complexes (HMEP)

• Humic Substances are Natural Organic Matter in Soil and Water Formed by Microbial Degradation of Plant Matter and Resistant to Further Degradation

Phytic Acid (Storage Form of P in Plant Tissue)

Apatite (Phosphate Minerals)

#### Figure 8-2. BAP% Comparison for Different Types of Advanced Nutrient Removal Processes



>50% Effluent P Recalcitrant in Most Cases

# Comparison of Biological P Removal v. Chemical P Removal

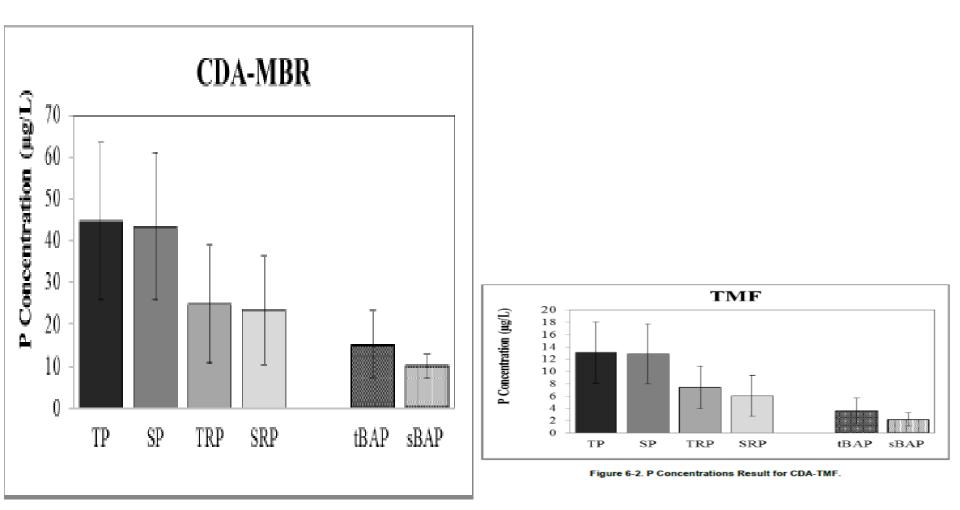


Figure 4-1. P Concentrations Result for MBR System in Coeur d'Alene.

tBAP<20 ug/L

tBAP<5 ug/L

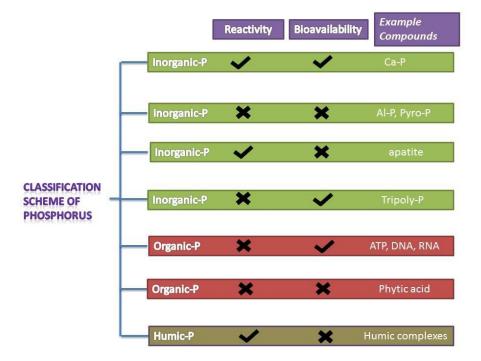
### **Bioassays of 21 P Compounds**

### Findings

- 17 Cases Did Not Fit Classic
   Assumption That SRP = Bioavailable
- Humic- (AI/Fe)-P Complexes
  - Alias as SRP. Not Bioavailable
- Phytic Acid Low BAP in Effluent
- Apatite Low BAP in Natural Waters

### Implications

- "Classic" Measurements Generally are SRP, TP, etc
  - WQ Model Setup Uses "Classic" Measurements as Input
- Simulation of BAP
  - Adapt WQ Model Setup to New Measurements/Classifications of P



### Conclusions

 Water Quality Modelers and Permit Writers Should Consider Importance of BAP

For Ultra Low Effluent P (<100 ug/L) Circumstances</li>

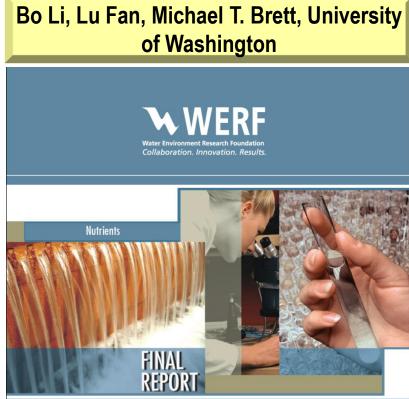
 Total Reactive Phosphorus (TRP) Strongly Associated with Bioavailable Phosphorus (BAP)

 $_{\odot}$  TRP Provides Fast, Simple and Conservative Estimate of BAP

 $_{\odot}$  TRP Gives Good Estimate of BAP without Bioassay Work

### "Mineralization Kinetics of Soluble Phosphorus and Soluble Organic Nitrogen in Advanced Nutrient Removal Effluents" (NUTR1R06p)

- Mineralization Kinetics for Phosphorus
  - Organic P Turned Into Inorganic
- Algal Bioassays
- Effluent Sampled
  - Spokane County
  - Post Falls
  - $_{\circ}$  Hayden
  - $_{\circ}$  Coeur d'Alene
  - Inland Empire Paper



Mineralization Kinetics of Soluble Phosphorus and Soluble Organic Nitrogen in Advanced Nutrient Removal Effluents



### Findings

- Discharger Specific Mineralization Rates for Dissolved Phosphorus
  - Biological Treatment, Membrane Bioreactor, Tertiary Membrane, Dual Sand Filtration
    - First Order Rate Kinetics That Can Be Used in Spokane River Model
      - » 1, 2 and 3 Pool Models
        - 2 and 3 Pool Models Correlated Better Than 1 Pool
    - Gamma Model
      - » 2 Parameter Continuous Probability Distribution
- Determined Bioavailability of Nitrogen and Phosphorus in Effluent from Nutrient Removal Facilities
- Characterizes Effluent Composition
  - P Composition from Bioavailable Phosphorus (BAP) Experiments
  - N Composition from Bioavailable Nitrogen (BAN) Experiments

### Conclusions

#### **Phosphorus**

- Bioassay Experiments Can Describe P Mineralization Rates
  - Easy Incorporation of First Order
     Decay Kinetics into Water Quality
     Models
    - Gamma Model May be Better
- 16% to 63% Dissolved P Recalcitrant

### <u>Nitrogen</u>

- Cell Growth Bioassay Not Suitable for Bioavailable N Analysis
  - $_{\odot}$  Dissolved N Uptake More Reliable
- 32% to 84% Dissolved N Recalcitrant

 Further Research Into Recalcitrant N and P Fractions and Potential Connection to Larger Molecular Sizes





### **Applications**

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### **Spokane River**

### Dissolved Oxygen Total <u>Maximum Daily Load (TMDL)</u>

- 8 mg/l Dissolved Oxygen Standard Not Met in Lake Spokane
  - Allowable 0.20 mg/l D.O.
     Depression
    - Washington Dissolved Oxygen Water Quality Standards Are Restrictive
      - » Standards allow for a small depression in natural dissolved oxygen levels for all human activities (<0.2 mg/L) if a waterbody is naturally lower than the state's numerical oxygen standard
- Very Low Effluent P Limits
  - TMDL WLA: Total Phosphorus 36 ug/L Idaho/42 ug/L Washington

### BAP Relevance?

- Significant For Ultra Low Effluent P (<100 ug/L) Circumstances</li>
  - o >50% Effluent P Recalcitrant in Most
    - Cases



Lake Spokane Algae Bloom. Source: Washington Department of Ecology October 2004 Draft TMDL

### **Nutrient Bioavailability and Permitting**

#### **Potential Applicability**

- Low N (<5 mg/L) and Low P Effluent (< 100 ug/L)</li>
  - $_{\odot}\,$  Refractory/Slowly Degradable N and P
    - Not Biodegradable in Treatment
    - Reduced Bioavailability (<50%)
- Effluent Limits Based on Bioavailability
  - Total Reactive P (TRP)?
    - Avoids Recalcitrant Constituents

Example Nitrogen NPDES Permit LOTT Budd Inlet Plant Effluent Limits • Total Inorganic Nitrogen (TIN) 2 mg/L

#### **Under Consideration**

- Onondaga Lake TMDL, Syracuse, NY
  - Onondaga County & NYDEC
- Spokane River DO TMDL
  - Spokane County (Washington Ecology)

Parameter	Seasonal Limit Applies March 1 to October 31 See notes f and g		
Carbonaceous Biochemical Oxygen Demand (5-day) (CBOD <sub>5</sub> )	133.4 pounds/day (lbs/day) average		
Total Phosphorus (as P) March 1 to Oct. 31	3.34 lbs/day average		
Total Ammonia (as NH3-N)	Seasonal Limit	Maximum Daily Limi	
For "season" of March 1 to March 31	1067.5 lbs/day average	16 mg/L	
For "season" of April 1 to May 31	66.7 lbs/day average	16 mg/L	
For "s Permit Footnote	· "gEuture add	8 mg/L	

For 's Permit Footnote: "<sup>g</sup>Future adjustments to the final effluent based on demonstrated pollutant equivalencies or non bioavailable P will be implemented as major modifications requiring public notice and comment"

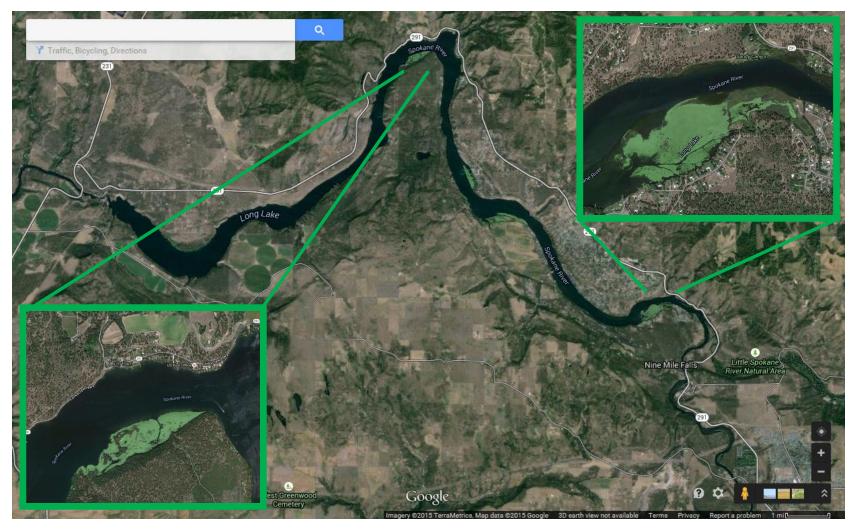
### **Bioavailability Considerations and Treatment Process** Design

- Influent Wastewater Characteristics
  - Potential Sources of Dissolved Organics and Recalcitrant Species
    - Industrial Loadings, etc.
- Treatment Process Selection
  - Single v. Multiple Stage Processes
  - Biological P Removal v. Chemical
    - Biosolids Land Application and P-Index Controlled Application Rates
- Generation of Recalcitrant Species within Treatment Processes
  - o Solids Processing and Anaerobic Digestion?
- Treatment for Removal of Recalcitrant Nutrient Species
  - ∘ Humic Substances and Humic-Metal-P Complexes (HMEP)?
    - Convert to Something That Can Be Removed in Treatment?
      - » Advanced Oxidation Processes (AOP) (UV + Peroxide) Coagulate & Filter?



#### **Bioavailability Considerations and Water Quality**

### Observations of Eutrophication



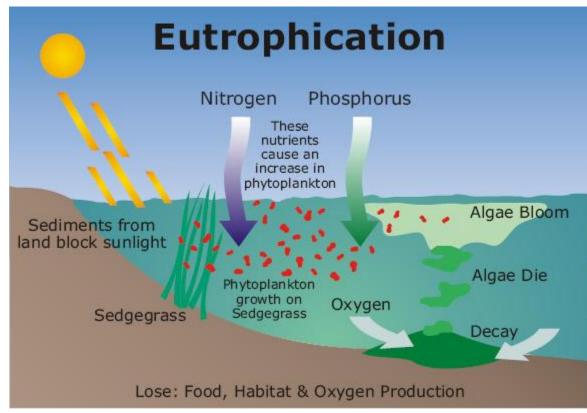
#### **Basis for Water Quality Modeling**

### Eutrophication Process

 $_{\odot}$  Nutrients (Nitrogen and Phosphorus) promote algae

blooms

 Algae die and consume oxygen



Source: Ohio Wesleyan University

### Water Quality Monitoring Data to Construct Models

- Collect samples and analyze in the laboratory
- Per Washington Ecology
  - $_{\odot}$  River and Stream Water Quality Monitoring
    - » <u>http://www.ecy.wa.gov/programs/eap/fw\_riv/rv\_main.html</u>
- Routinely measured indicators are:
  - ammonia (SM4500NH3H)
  - nitrate plus nitrite (SM4500NO3I)
  - nitrogen, total (SM4500NB)
  - phosphorus, soluble reactive (SM4500PG)
  - phosphorus, total (SM4500PH)
  - conductivity (SM2510B)
  - suspended solids (SM2540D)
  - fecal coliform bacteria (SM9222D)
  - oxygen (SM4500OC)
  - temperature (thermistor)
  - flow (at most stations)
  - pH (EPA150.1)
  - turbidity (SM2130)
  - metals (bimonthly, at a few stations) (EPA200.8, 245.7)

#### Water Quality Models

### Water Quality Models are Tools to Relate the Measured Parameters

 $_{\odot}$  Example: CE-QUAL-W2

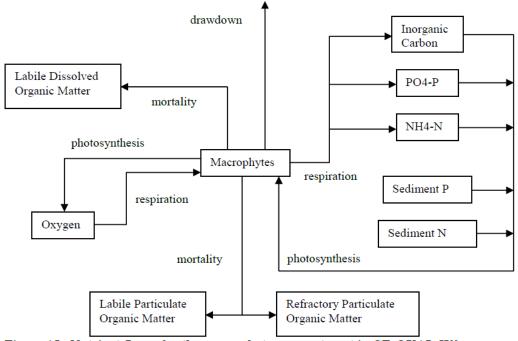


Figure 15. Nutrient fluxes for the macrophyte compartment in CE-QUAL-W2.

## Summary of Receiving Water Model Capabilities to Simulate Nutrient Species and Refractory Compounds

	Model Label or Name Acronym	Current Capabilities	<b>Opportunities for Enhanced</b> <b>Application and Potential Refinements</b>		
	Simplistic Models				
Ma	lass Balance	Data Assessment Technique. Incapable of Simulating Nutrient Species.	Limited Potential.		
En	mpirical	Data Assessment Technique. Incapable of Simulating Nutrient Species.	Limited Potential.		
	Mechanistic Models				
AC	QUATOX	Capable of Nutrient Species Simulation.	Provide Data on Refractory Compounds.		
		Bioavailability reflected with input of SRP and refractory organic phosphorus.	Improve documentation and explanation in the model interface and user manual about nutrient species options. Provide multiple options for nutrient speciation and coefficient conversion rates.		
CE	E-QUAL-ICM	Nutrient Speciation and Refractory Compounds.	Provide Data on Refractory Compounds.		
	E-QUAL-RIV1	Capable of Nutrient Species Simulation.	Provide Data on Nutrient Species.		
CE-QUAL-W2	E-QUAL-W2	Nutrient Speciation and Refractory Compounds.	Provide Data on Refractory Compounds.	Need Data	
		Bioavailability reflected with input of SRP and organic phosphorus associated with BOD.	Model can divide point source phosphorus loads into as many unavailable compartments as desired. Set an upper bound mineralization rate for weakly bioavailable compounds. Requires monitoring data to calibrate mineralization rates.		
DE	ELWAQ	Nutrient Speciation and Refractory Compounds.	Provide Data on Refractory Compounds.		
EF	FDC/HEM-3D	Nutrient Speciation and Refractory Compounds.	Provide Data on Refractory Compounds.		
M	IKE	Capable of Nutrient Species Simulation.	Provide Data on Nutrient Species.		
RC	CA	Nutrient Speciation and Refractory Compounds.	Provide Data on Refractory Compounds.		
QI	UAL2K	Capable of Nutrient Species Simulation.	Provide Data on Nutrient Species.		
		Bioavailability reflected with input of Soluble Reactive Phosphorus (SRP) only.	Add options for nutrient speciation with point source input specification. Add options for mineralization with rates input specification.		
W	ASP7	Capable of Nutrient Species Simulation.	Provide Data on Nutrient Species.		

#### Incorporate into CE-QUAL-W2 Water Quality Model

- "The model can track multiple organic matter fractions and nutrient fractions associated with the organic matter fractions separating these into both particulate and dissolved and refractory and labile. For example, one can have a CBOD group and the CBOD-N and **CBOD-P** associated with one discharger and these could be refractory – the same discharger could also use another CBOD group that has the same C, N, and P fractions that are labile. Similarly for dissolved and particulate versions of each of these. Since the model can have as many CBOD groups as one wants, then there is no limit to the amount of slicing and dicing and tracking one can do for the speciation of organic matter!" - Scott Wells, Portland State University, CE-QUAL-W2 model developer
- "....there is no limit to the amount of slicing and dicing and tracking one can do for the speciation of organic matter!"

#### Water Quality Model Inputs

### Inputs Generally Align with Parameters That are

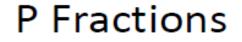
Measured

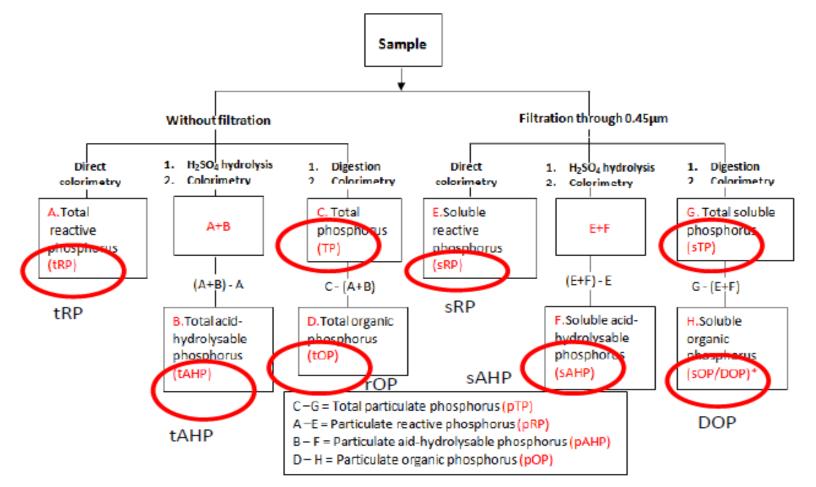
CIN CON			CINBRC		CINBRC CINBRC CINBRC CINBRC
TDS	ON	ON	ON		Total dissolved solids or salinity
TRACER	ON	ON	ON		Generic constituent 1 - tracer
AGE	OFF	OFF	OFF		Generic constituenT 2 - residence time
COL1	ON	ON	ON		Generic constituent 3 - coliform group 1
COL2	ON	ON	ON	ON	Generic constituent 4 - coliform group 2
ISS1	ON	ON	ON	ON	Inorganic suspended solids group 1
ISS2	ON	ON	ON	ON	Inorganic suspended solids group 2
ISS3	ON	ON	ON	ON	Inorganic suspended solids group 3
PO4	ON	ON	ON	ON	Inorganic dissolved phosphorus
NH4	ON	ON	ON	ON	Ammonium
NO3	ON	ON	ON	ON	Nitrate-nitrite
DSI	ON	ON	ON	ON	Dissolved silica
PSI	ON	ON	ON	ON	Particulate biogenic silica
FE	ON	ON	ON	ON	Iron
LDOM	ON	ON	ON	ON	Labile dissolved organic matter
RDOM	ON	ON	ON	ON	Refractory dissolved organic matter
LPOM	ON	ON	ON		Labile particulate organic matter
RPOM	ON	ON	ON	ON	Refractory particulate organic matter
CBOD1	ON	ON	ON	ON	Carbonaceous BOD group 1
CBOD2	ON	ON	ON	ON	Carbonaceous BOD group 2
CBOD3	ON	ON	ON	ON	Carbonaceous BOD group 3
CBOD1-P	ON	ON	ON	ON	Carbonaceous BOD-P group 1
CBOD2-P	ON	ON	ON	ON	Carbonaceous BOD-P group 2
CBOD3-P	ON	ON	ON	ON	Carbonaceous BOD-P group 3
CBOD1-N	ON	ON	ON	ON	Carbonaceous BOD-N group 1
CBOD2-N	ON	ON	ON	ON	Carbonaceous BOD-N group 2
CBOD3-N	ON	ON	ON	ON	Carbonaceous BOD-N group 3
ALG1	ON	ON	ON	ON	Algal group 1
ALG2	ON	ON	ON	ON	Algal group 2
ALG3	ON	ON	ON	ON	Algal group 3
DO	ON	ON	ON	ON	Dissolved oxygen
TIC	ON	ON	ON	ON	Total inorganic carbon mg/l as C
ALK	ON	ON	ON	ON	Alkalinity mg/l as CaCO3
Z001	ON	ON	ON		Zooplankton
LDOM P	ON	ON	ON	ON	Total P in labile dissolved organic matter
RDOM P	ON	ON	ON	ON	Total P in refractory dissolved organic matter
LPOM P	ON	ON	ON		Total P in labile particulate organic matter
RPOM P	ON	ON	ON	ON	Total P in refractory particulate org matter
LDOM N	ON	ON	ON	ON	Total N in labile dissolved organic matter
RDOM N	ON	ON	ON	ON	Total N in refractory dissolved organic matter
LPOM N	ON	ON	ON	ON	Total N in labile particulate organic matter
RPOM N	ON	ON	ON	ON	Total N in refractory particulate org matter

#### **Model Data Needs**

- Model entries for ambient and effluent nutrient loadings can reflect nutrient speciation and refractory organic compounds, providing that the data are available to characterize effluent discharges and ambient waters
- BAP Studies Directly Inform WQ Model Needs

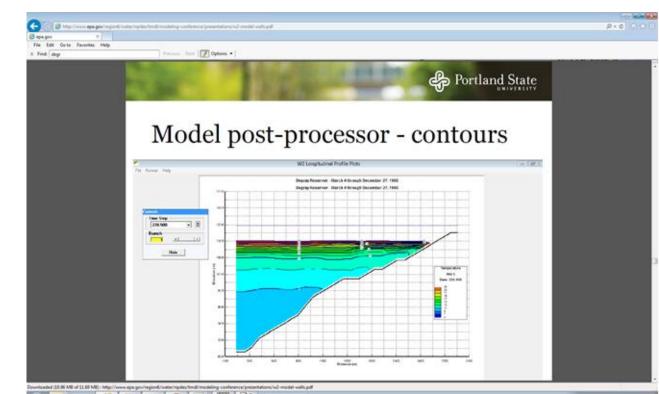
#### **P** Speciation Terminology





### Water Quality Model Test of Sensitivity to BAP Inputs

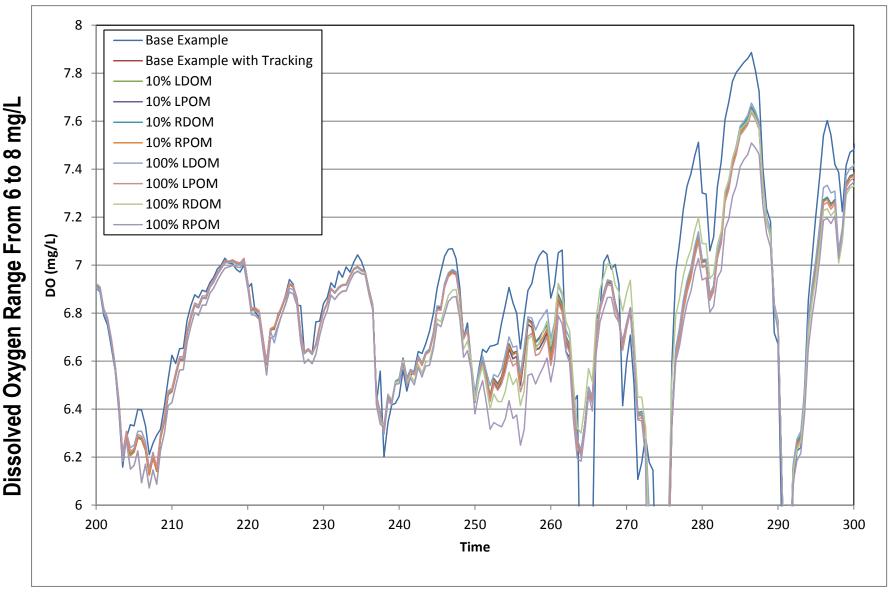
- CE-QUAL-W2 Example Model
  - $_{\rm O}$  DeGray Reservoir, AR
    - ~15 Miles Long
    - ~60 Feet Dee
    - Model Period ~1 year



#### Model Scenarios to Explore Dissolved Oxygen Sensitivity to Changes in P Bioavailability

- Difference in dissolved oxygen in a reservoir when 10and 100-percent of the SRP is entered as
  - $_{\circ}$  Labile dissolved
  - Labile particulate
  - $_{\odot}$  Refractory dissolved
  - $_{\circ}$  Refractory particulate

#### Example Dissolved Oxygen Sensitivity Results to Changes in P Bioavailability in Lowermost Segment of Reservoir



Time: Julian day 200 (July 19<sup>th</sup>) to Julian day 300 (October 27<sup>th</sup>)

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